$$\operatorname{CO}_2$$
 CO_2 CO_2 CO_2 CO_2 CO_2 CO_2 Eu^{3+} Eu^{3+} Eu^{3+}

4

5

6

Figure 1

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Ytterbium Chelate

Sodium salicylate

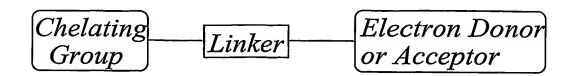
Thianthrene

Diphenyloxazole

Carbostyryl 124

Figure 2

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$$C = C$$
 $C = C$
 $C = C$
 $C = C$

Figure 3

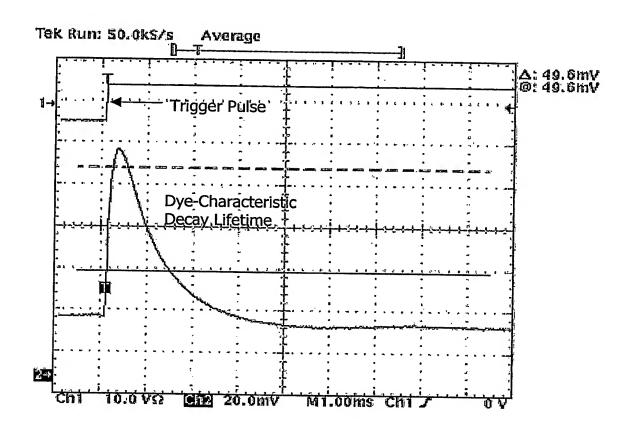


Figure 4

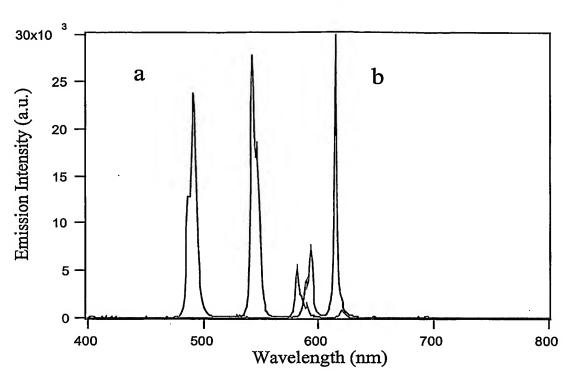


Figure 5

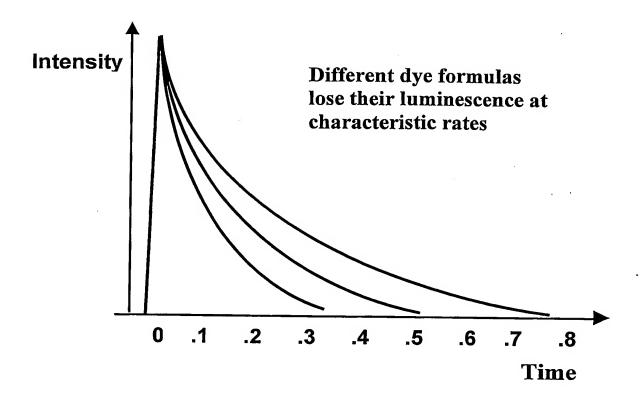


Figure 6

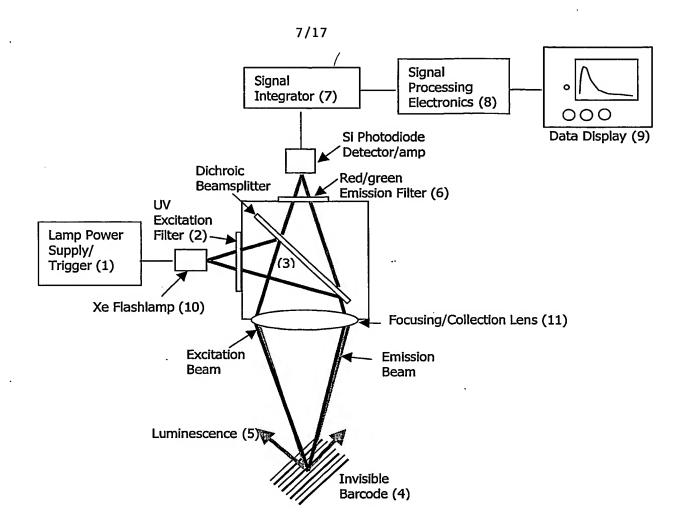


Figure 7

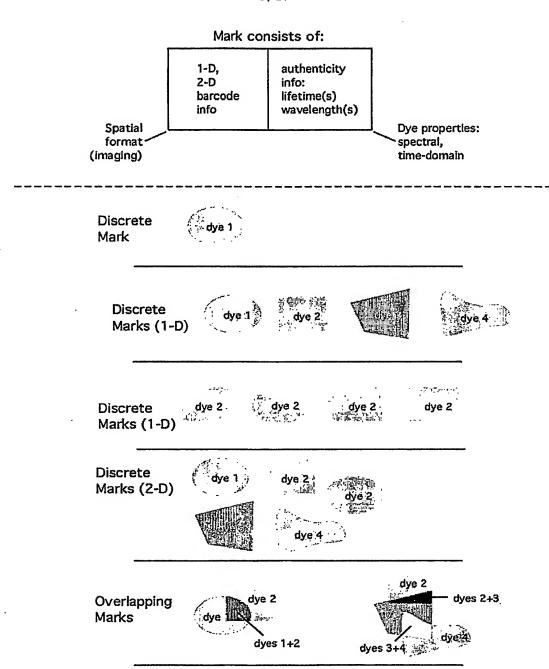


Figure 8

System¹Operation

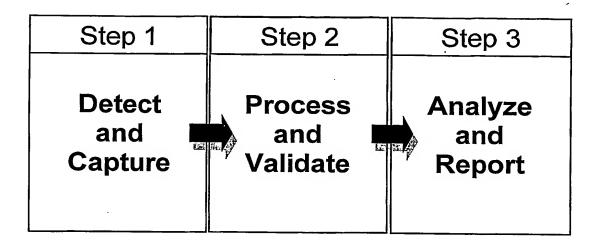


Figure 9A

Step 2 Process and Validate

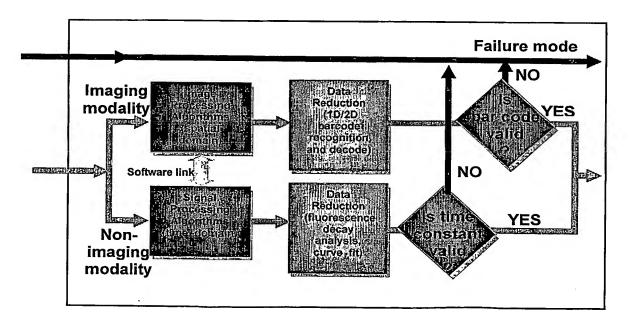


Figure 9B

Step 3 Analyze and Report

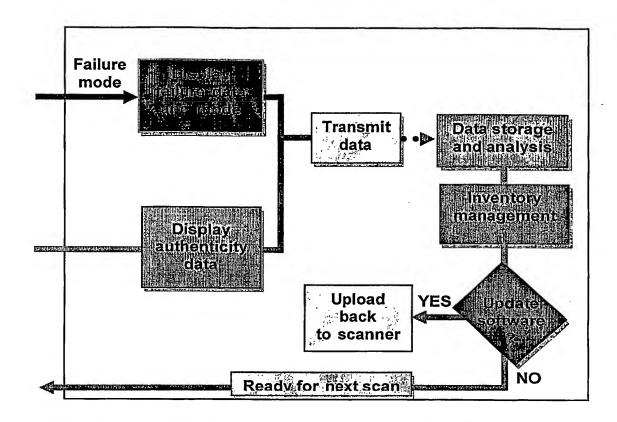


Figure 10

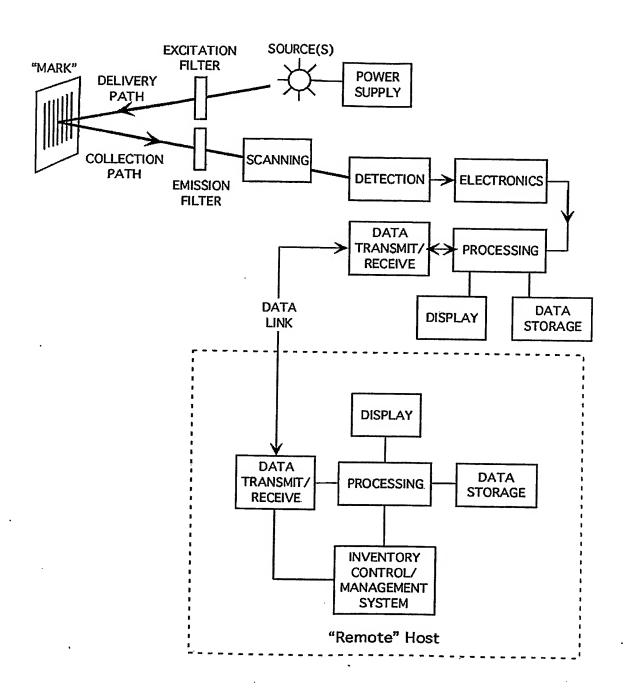
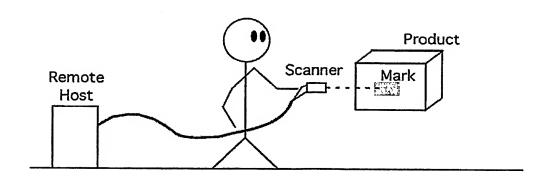
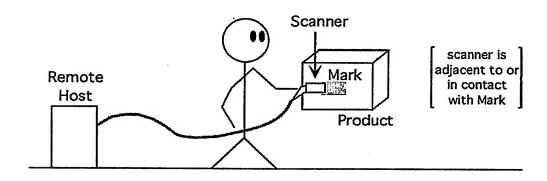


Figure 11





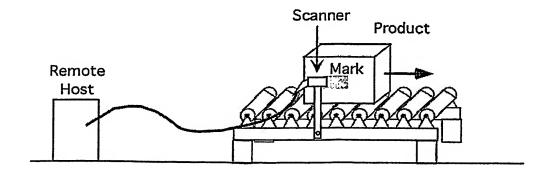


Figure 12

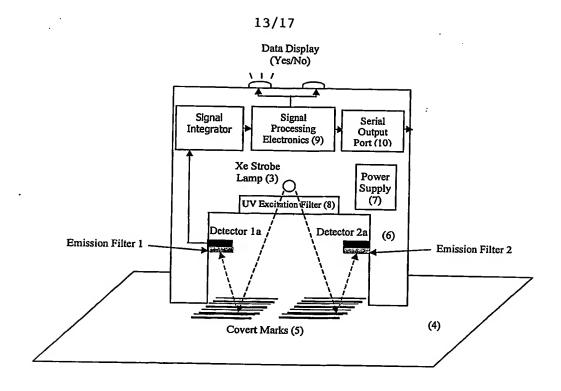
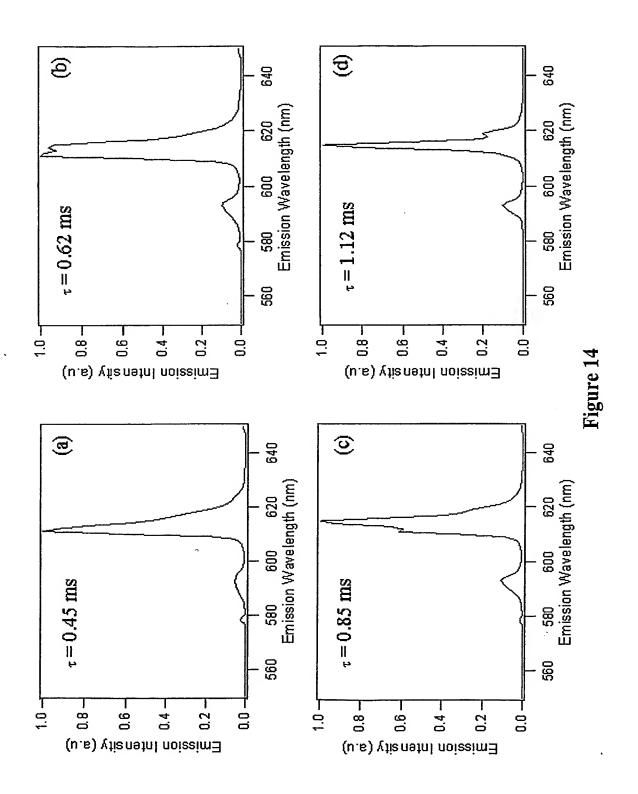


Figure 13

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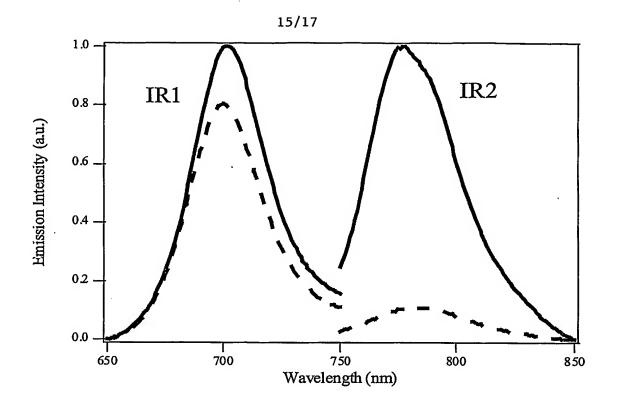


Figure 15

Table 1 Absorption and luminescence parameters for europium chelates ^a

Chelates	λ _{max (abs)}	$\epsilon_{ ext{max}}$	λ _{max (lum)}	$ au_{ m lum}$
1	308 nm	17400	615 nm	0.97 ms
2	326 nm	16600	615 nm	1.02 ms
3	317 nm	17200	614 nm	1.01 ms
4	337 nm	19300	615 nm	0.97 ms
5	339 nm	26000	614 nm	0.36 ms
6	352 nm	24200	615 nm	0.43 ms

a, Aerated water solutions, pH 8.5; [chelate] = 40 μ M; λ_{exc} @337 nm for lifetime τ_{lum} measurements; τ_{lum} recorded by monitoring emission at 615 nm.

Table 2 Luminescence lifetime data for chelates

Chelate		Modifiers ^a		
	None	Imidazole	4-MI	IDA
1	0.97 ms	1.19 ms	1.22 ms	0.45 ms
2	1.02 ms	1.21 ms	1.37 ms	0.48 ms
3	1.01 ms	1.23 ms	1.19 ms	0.46 ms
4	0.97 ms	1.27 ms	2.03 ms	0.60 ms
5	0.36 ms	0.65 ms	0.54 ms	0.50 ms

a, Modifiers at 1.0 mM concentration; Imidazole, 4-methylimidazole (4-MI), and iminodiacetic acid (IDA), were added to an application medium of 200 μ M chelate in water (pH9.0) and applied to plain paper; lifetimes (units of ms) from exponential luminescence decay curves ($\lambda_{exc} = 337$ nm; $\lambda_{lum} = 615$).

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